CHEM 1204 GENERAL CHEMISTRY II LABORATORY  

1 credit

Fall 2015

2002-2004 Undergraduate Studies Bulletin Description:
Practical applications of the fundamental laws, theories, and principles of chemistry through problem solving and laboratory experiments.

Prerequisites:
CHEM 1201 General Chemistry I (lect), CHEM 1203 General Chemistry I Laboratory, Elementary algebra

Corequisites:
CHEM 1202 General Chemistry I (Lecture)

Required Textbook: General Chemistry Laboratory Manual by Dr. Christopher Devine (Professor Emeritus), Dr. James Dougherty, Dr. Ish Kumar, Dr. Mihaela Leonida, and Dr. Arthur Murphy) 

Fall 2014 Edition (available at the University Bookstore).

Course objectives and outcomes:

Objective 1: To promote proper laboratory practices and report preparation 
(Programmatic Outcome #1,2,3,4,5)
Outcome 1.1: Know location of safety equipment, be familiar with emergency procedures and proper laboratory attire.

Outcome 1.2: Understand laboratory report format and grading criteria.

Outcome 1.3: Use Microcomputers to assist in report preparation.

Outcome 1.4: Learn about data limitations and experimental uncertainties.

Objective 2: Become proficient at handling chemicals and using laboratory equipment. 
(Programmatic Outcome #1,2,3,4,5)
Outcome 2.1: Be trained in handling acids, bases, flammable and toxic substances.

Outcome 2.2: Be trained in proper use of balances, burets, eudiometers, Bunsen burners, pychnometers, various types of glassware, stands and clamps.

Objective 3: Reinforce material presented in the General Chemistry I recitation course. 
(Programmatic Outcome #1,2,3,4,5)
Outcome 3.1: Perform experiments on temperature and density measurements.

Outcome 3.2: Perform experiments illustrating stoichiometry and chemical reactions.
**Outcome 3.3:** Perform experiments related to gas laws, thermochemical concepts, and molecular structure.

**General “Outcomes” Remarks:** Students who successfully complete this course should have deepened their knowledge of the theoretical material discussed in General Chemistry I lecture. The topics discussed are all fundamental, and they should serve as a basis for future studies in biology, chemistry, and other sciences. This course is a prerequisite for General Chemistry II laboratory (CHEM 1204).

**Overview**
Since there are multiple sections of General Chemistry II Laboratory, the following syllabus should be regarded as a generic syllabus. All laboratory sections perform the same experiments, but there may be Instructor specific differences with regard to certain policies such as grading procedures etc. During your first laboratory period, you will receive additional instructions regarding the exact policies being used.

**Tentative Laboratory Schedule (Fall - 2015)**

<table>
<thead>
<tr>
<th>Week #</th>
<th>Experiment</th>
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<tbody>
<tr>
<td>2</td>
<td>Exp. #15 Molar Mass Determination by Freezing Point Depression.</td>
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<tr>
<td>3</td>
<td>Exp. #16 Chemical Kinetics I: The Method of Initial Rates.</td>
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<tr>
<td>4</td>
<td>Exp. #17 Chemical Kinetics II: Integrated Rate Laws</td>
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<tr>
<td>5</td>
<td>Exp. #18 Determination of the Equilibrium Constant of a Chemical Reaction</td>
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<tr>
<td>6</td>
<td>Exp. #19 Titration of a Polyprotic Acids and Determination of Acid Dissociation Constants.</td>
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<tr>
<td>7</td>
<td>Exp. #20 Solubility Products, Common Ion Effect, Hardness of water</td>
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<tr>
<td>8</td>
<td>Exp. #21 Analysis of Group III ions</td>
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<tr>
<td>9</td>
<td>Exp. #22 A Redox Titration</td>
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<tr>
<td>10</td>
<td>Exp. #23 Electrochemical Cells.</td>
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<tr>
<td>11</td>
<td>Exp. #24 Ion Exchange Chromatography</td>
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<tr>
<td>12</td>
<td>Exp. #25 Spectroscopy of Complex Ions.</td>
</tr>
<tr>
<td>13</td>
<td>Exp. #26 Organic Chemistry and the Synthesis of Esters.</td>
</tr>
<tr>
<td>14</td>
<td>Check out. Lab final exam.</td>
</tr>
</tbody>
</table>

The last day for withdrawing from the course with a grade of "W" is October 30.

During the first lab session, your instructor will explain the laboratory report format to be used, the policy regarding lab quizzes, and the general grading policy used for the course. Unless otherwise stated, laboratory reports are due one week after the completion of the experiment. Late lab reports will not be accepted. No make-ups of missed labs are allowed. Students are generally allowed only one excused absence from lab.

Your instructor may also opt to supplement some of the experiments listed above with (a) video tapes...
illustrating various chemistry principles, (b) computer exercises, or (c) demonstrations.

**Safety Procedures and Policies:**
During the first laboratory period, your instructor will explain the safety procedures and policies that are to be followed in this course. All students are expected to follow these procedures at all times. Appropriate laboratory attire is required. No student will be permitted into laboratories wearing shorts, halter-tops, open toed sandals, undershirts, tank tops or any other inappropriate attire. All students are to purchase a white laboratory coat which can be used for any Biology or Chemistry class which requires a lab - non-majors or majors.

Below you will find URL’s for FDU’s Academic Integrity policy as well as FDU’s Accommodations Policy, for students with learning disabilities, or requiring assistance.

**Academic Integrity Policy**

http://view2.fdu.edu/academics/university-college/school-of-humanities/writing-program-guide/academic-integrity-policy/

**Reasonable accommodations Policy**

http://view2.fdu.edu/metropolitan-campus/disability-support-services/